Code No: 121AE

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD B.Tech I Year Examinations, October/November - 2020 ENGINEERING CHEMISTRY

(Common to CE, EEE, ME, ECE, CSE, IT, ETM, AE, MIE, PTM)

Time: 2 hours Max. Marks: 75

Answer any five questions All questions carry equal marks

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- 1.a) Explain the advantages and applications of methanol oxygen fuel cell.
- b) How do you determine P^H of a solution using glass electrodes?

[7+8]

- 2.a) Discuss the factors effecting corrosion based on the nature of metal.
 - b) What is specific, equivalent and molar conductance of electrolyte? Derive expression.

[7+8]

- 3.a) Differentiate thermosetting and thermoplastic polymers.
 - b) Give the preparation, properties and engineering applications of Thiokol rubber.
 - c) Give an account on preparation and applications of poly lactic acid.

[5+5+5]

- 4.a) Explain compression molding of plastics.
 - b) What are the advantages of vulcanization? Explain the process.
 - c) How nano materials are prepared by sol-gel method?

[5+5+5]

- 5.a) What are scales and shuges. How are they caused and how they are removed from boilers.
 - b) Give an account on caustic embrittlement with chemical reaction.
 - c) Explain calgon and phosphate conditioning.

[5+5+5]

- 6.a) Why hardness of water is expressed in CaCO 3 equivalents. Give the causes and effects of hard water.
 - b) Explain zeolite process for purification of water.
 - c) Give the steps involve in the treatment of potable water.

[5+5+5]

- 7.a) Explain about fixed bed catalytic cracking.
 - b) How is coal analyzed by proximate analysis?
 - c) What is LCV and HCV? How are they related?

[5+5+5]

- 8.a) Explain the phase diagram for water system.
 - b) Give the application of colloids in industry.
 - c) Explain freundlichs adsorption theory. How does it differs from Langmuir's theory?

[5+5+5]

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